

**Mucoadhesive Film Forming Spray for Buccal Drug Delivery: A Review**

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**Abstract:**

Film-forming sprays provide a number of advantages over conventional topical treatments, including equal medication distribution and dosing, increased bioavailability (increase local drug concentration), and less irritability

(Dosing frequency). Polymers and excipients that improve the characteristics of preparations and increase the stability of active substances are the building blocks of mucoadhesive film-forming sprays. Films made from diverse combinations of polymer and excipient exhibited a wide range of characteristics. This review examines the many types of polymers and excipients, the different types of sprayers, the different evaluations, as well as the essential criteria that are involved in defining the sprayability and film properties. This comes to the conclusion that natural and synthetic polymers with viscoelastic properties can both be employed to optimize the administration of buccal drugs.

**Key words:** Film-Forming spray, Polymer, Buccal Drug Delivery System**رذاذ تشكيل غشاء مخاطي لتوصيل الأدوية الشدقية: مراجعة**

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**الخلاصة:**

عند مقارنة البخاخات المكونة للغشاء بالعلاجات الموضعية التقليدية، فإن البخاخات المكونة للغشاء لها العديد من الفوائد لأنها يمكن أن توفر توزيعاً موحداً للأدوية والجرعات، وتوافراً حيوياً أكبر، ونسبة أقل من التهيج. تتكون البخاخات اللاصقة المخاطية المكونة للغشاء من بوليمرات وسواغات تعمل على تحسين خصائص المستحضرات وتزيد من ثبات المكونات النشطة، وسوف ينتج كل بوليمر وسواغ أغشية ذات خصائص مميزة .  
يفحص هذا الاستعراض أنواعاً عديدة من البوليمرات والسواغات ، وأنواع مختلفة من المرشحات ، والتقييمات المختلفة ، فضلاً عن المعايير الأساسية التي يتم تضمينها في تحديد قابلية الرش وخصائص الفيلم. يأتي هذا إلى استنتاج مفاده أنه يمكن استخدام البوليمرات الطبيعية والأصطناعية ذات الخصائص اللزجة المرنة لتحسين إدارة الأدوية الشدقية.  
**الكلمات المفتاحية:** البخاخات المكونة للغشاء، بوليمر، نظام توصيل الأدوية الشدقية.

**Introduction**

Film dosage forms have acquired popularity in the pharmaceutical industry, patient-friendly, and convenient products. Orally dissolving films have just recently been brought to attention due to the higher mechanical properties that they possess [1].

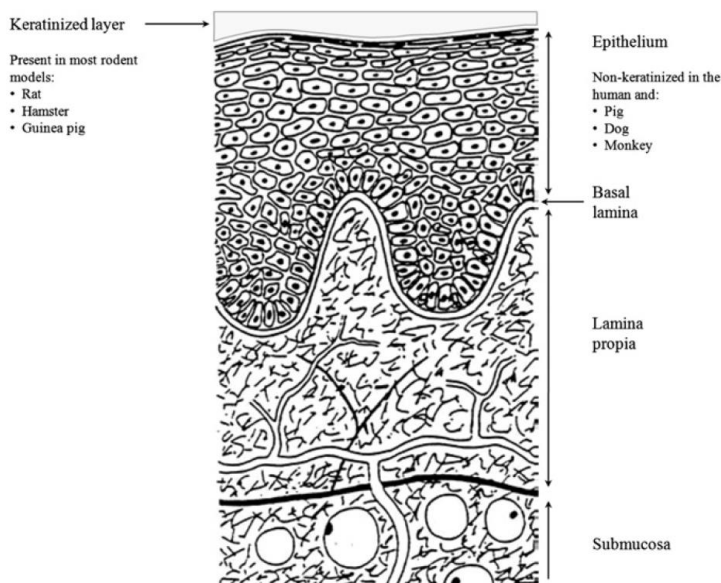
When compared to pills, mucoadhesive buccal films have a higher patient compliance because of their more manageable size and thickness [2]. There are different mucoadhesive buccal films that have been developed to administer medications locally in the mouth cavity in

order to treat fungal diseases such as oral candidiasis [3].

**Oral Mucosa Physicochemical Properties:**

About 25% of the oral cavity is made up of keratinized gingival and hard palate and

about 60% is covered by not keratinized mucosa lining the inside of the cheeks, the floor of the mouth and the tongue [4][5]. Figure 1. Showed that the epithelium is connected to a lamina propria, separated by a basal lamina



**Figure(1): Buccal mucosa cross-sectionb[6]**

The lamina propria and lining mucosa regions mainly serve as mechanical supports but do not pose a significant obstruction to the entry of active substances [7]. The blood vessels that empty into the internal jugular vein, as well as the lingual, facial, and retromandibular veins, are located in the connective tissue [8]. The main advantage of buccal delivery over oral administration is that the drug is absorbed directly into the portal vein, skipping the stomach and its related conditions (such as enzyme concentrations, first pass effect, and the gastric pH). As a result of the ease with which a given drug molecule can be transported after it enters connective tissue, the stratified epithelium's entire thickness serves as the penetration barrier [9].

**Mucoadhesion and mechanical properties of buccal mucosa**

Mucoadhesion is the interaction between the buccal mucosa, (which is a biological membrane that lines the buccal cavity) and polymers of the buccal drug delivery system (BDDS). In general, the adherence between a biological and synthetic surface is referred to as bioadhesion. A mucosal membrane's specific interaction with a synthetic surface is known as mucoadhesion [10]. Various theories explained this observation like wetting, electronic, adsorption, diffusion, mechanical, and fracture. All these hypotheses describe the influencers on mucoadhesion for example the bonds that hold mucoadhesive polymers and mucins together [11].

**Film-Forming Spray Mechanism**

A film forming spray (FFS) is a medication delivery system that sprays a solution or suspension onto a target treating site, where it contacts the site and uses the polymer as matrices to build a film [12]. Following the formation of the film, the drug will be released gradually similarly to a patch [13]. Naturally, this makes it much easier for drugs to reach their target tissues (long contact time with the buccal tissue).

Drug doses in a film-forming spray may be adjusted for systemic or local effects by

changing the amount of solution per each spray. Additionally, an FFS offers a good spread and an even distribution of medications. The simplicity of the treatment can also help in increasing patient compliance [14,15]. It is simple to clean the thin film with water in contrast to gels, ointment, and other similar substances (that have viscous consistency when used), the film has a smooth, elastic consistency, boosts patient comfort during activities [16][17]. Marketed oral spray examples are shown in Table 1.

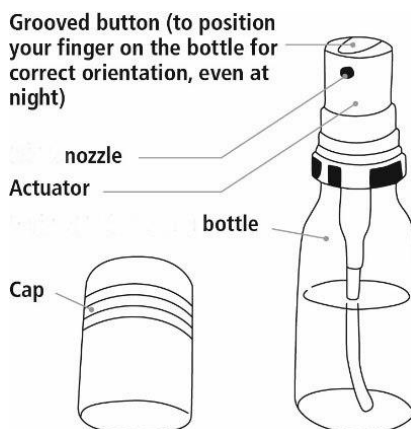
**Table (1): Marketed oral sprays[18]**

| Product                   | Content                                 | Commercial name | Special technology or properties |
|---------------------------|---|-----------------|----------------------------------|
| Buccal Mist               | Mouth spray of Insulin                  | Oral-lyn™ spray | RapidMist™ spray dose Technology |
| Sublingual Spray solution | Sublingual spray of Glyceryl trinitrate | Glytrin Spray®  | Metered-dose spray               |
| Throat spray              | Throat spray of Flurbiprofen            | Benactiv®       | Metered-dose spray               |
| Sublingual                | Isosorbide dinitrate                    | Isocard spray   | Metered-dose aerosol.            |

**Parts of Film-Forming Sprays and Types**

Nozzle design, aperture size, spray pressure, and liquid type all have

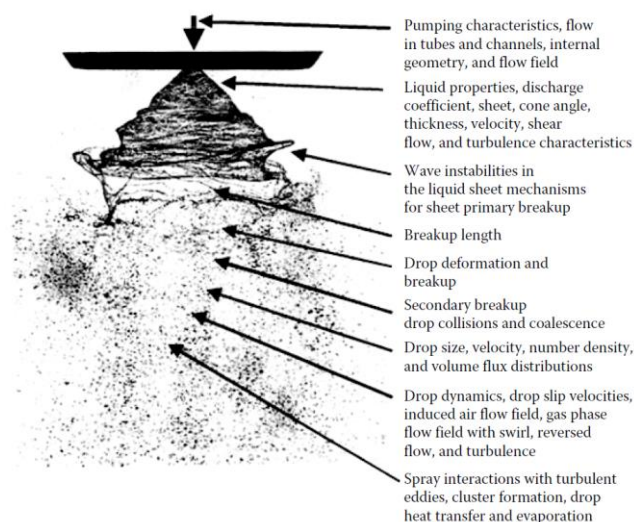
substantial impacts on FFS sprayability [19,20]. A regular spray pump in a can or bottle is the most convenient and efficient method of dispensing the product. A preferred valve actuator is a mechanical breakup actuator [21] (Figure 2).



**Figure (2): Parts of buccal spray [22]**

The spray pressure, liquid characteristics, and nozzle type also have a role in determining the size of the resulting droplets. However, a single nozzle is capable of producing droplets of varying sizes, which is not always necessarily

uniform, as shown in Figure 3. The volume sprayed per area, the uniformity of application, the coverage, and the possible danger of drift are all affected by the spray nozzles that are used [23].



**Figure (3):Shape and characteristics of the spray** [24]

Aerosol systems which use the pressure of compressed or liquefied gas to force the contents of the container. Metered dose sprays, which do not use propellants rely on solvents and polymers to deliver a controlled quantity of medicine per each actuation, have recently replaced the propellant based sprays [25].

The propellant provides the force to expel the product from the container in the desirable form (spray, semisolid, foamy form). The propellant can be as a liquefied gas or mixing of liquefied gases, moreover, it can be used as the product concentrate's

solvent. An example of propellant used in an oral formulation is Chlorofluorocarbon (CFC) [26].

Metered dose spray was invented by a Victorian College of Pharmacy (Australia) and commercialized by Acrux Limited [27].

It is a spray apparatus that can calibrate the amount of spray as shown in Figure 4.

This device is often used for transdermal or transmucosal administration of preparations destined for the systemic compartment. In assessing a film forming spray, volume of spray is crucial to drug dose [28].



**Figure (4): Metered dose spray** <sup>[28]</sup>

The characteristics of the spray film, such as its viscoelasticity, in situ gelation, pH sensitivity, and temperature sensitivity, are essential to determining what considerations should be made when selecting polymeric materials, surfactants, and other optimized formulations. These characteristics are important because they help determine what factors should be taken into account <sup>[29]</sup>. Research on the components applied through buccal spray was included in the compiled literature as in the formation of Lidocaine Hydrochloride buccal spray where two different types of carbopol (934 and 940) were employed as film forming polymers and both types of carbopol and Lidocaine Hydrochloride precipitate when combined. Precipitate formation was avoided and preparation stability was increased by using additional polymer like xanthan gum, also they used propylparaben as a preservative <sup>[30]</sup>.

### **Components of Film Forming Spray Film Forming Spray Polymers**

A summary of the many polymers that are put to use in the manufacture of buccal films in their various concentrations. There are natural polymers (such as hydroxypropyl methylcellulose HPMC), semi-synthetic polymers (such as ethyl cellulose), and synthetic polymers (such as polyvinyl pyrrolidone) <sup>[31]</sup>. Also, there are non-ionic polymers (such as hydroxyl ethyl cellulose HEC), anionic polymers (such as sodium carboxymethyl cellulose SCMC) as well as a cationic polymer (such as chitosan) <sup>[32,33]</sup>.

### **Film-Forming Spray Excipients**

To enhance the preparation's quality and therapeutic effectiveness, excipients other than polymers are also included such as the plasticizer which keeps the film elastic during film formation and stops it from cracking, increasing drug permeability while ensuring the stability of productivity <sup>[34]</sup>, as shown in Table 2.

**Table (2): List of excipients used for buccal film-forming spray**

| Class of excipient | Examples  | Function                            |
|--------------------|---|-------------------------------------|
| Surfactants        | Sodium dodecyl sulfate <sup>[35]</sup>  | Permeation enhancer                 |
|                    | Sodium lauryl sulfate <sup>[36]</sup>   |                                     |
|                    | Polyethylene glycol 200 <sup>[37]</sup>   | Plasticizer                         |
|                    | Polyethylene glycol 400 <sup>[28]</sup>   |                                     |
| Bile salts         | Sodium glycocholate <sup>[38,39]</sup><br>Sodium taurocholate, sodium glycol deoxycholate, and sodium tauro deoxycholate <sup>[38]</sup><br>Sodium deoxycholate <sup>[40]</sup> | Permeation enhancer                 |
| Fatty acids        | Oleic acid <sup>[41-43]</sup><br>Eicosa pentaenoic acid <sup>[44]</sup> and or docosa hexaenoic acid <sup>[43]</sup>  | Permeation enhancer                 |
| Fatty acid ester   | Propylene glycol (PG) <sup>[45]</sup>   | Plasticizer and permeation enhancer |
| Alcohol            | Ethanol <sup>[46,47]</sup>  | Permeation enhancer                 |

Propylene glycol (PG) and polyethylene glycol (PEG) play a part in enhancing the penetration of antifungal medications. In addition to its function as a plasticizer, PG also serves as a solubilizer, which helps deliver medications <sup>[48]</sup>. The concentration of PG is important since it has a large impact on the film forming liquid's viscosity <sup>[49]</sup>.

Polyethylene glycol 400 can be used to increase the volume of a film-forming solution of each puff. With higher polyethylene glycol 400 concentrations, the amount of each spray rises. Increasing polyethylene glycol 400 levels also increase the covered spray area <sup>[50]</sup>. Because PEG is a nonvolatile solvent <sup>[51]</sup>, so the vapor pressure decreases <sup>[52]</sup>. Permeation enhancers also involved in film forming spray (FFS) that enhanced the paracellular penetration of the drug through the mucosal membrane such as; sodium glycocholate, oleic acid, lauric acid, and propylene glycol <sup>[53]</sup>.

### Solvents

The FFS employs both volatile and not volatile solvents. To maintain a steady rate of drying for the film. Films which have fast drying time make difficulty for the drug to enter and release from the

produced hard films. In most cases, the active substance is dissolved within the solvent until it comes close to being saturated, which speeds up the drying process of the film <sup>[54]</sup>.

### Evaluation of Film-Forming Sprays pH

The pH value is evaluated and changed to maximize the active ingredient's stability or suitability for the application area. The content and volume of saliva has a pH range of 5.9 to 7.3 <sup>[55]</sup>. In order to avoid irritation and changes in the physiological state, the preparation's pH has been adjusted <sup>[56]</sup>. As shown in pH of Rizatriptan benzoate films was in the range of 6.54 to 6.98 for all formulations indicating that no irritation is expected <sup>(57)</sup>. Also in Tizanidine hydrochloride buccal formulations, pH results ranged from 5.7 to 6.8 <sup>[58]</sup>.

### Viscosity

The viscosity of the polymer will vary depending on its type and dosage. The film-forming solution's viscosity will influence its sprayability, making it a crucial factor, especially in MDS <sup>(59)</sup>. The spray's coverage can be adjusted by adjusting the concentration of the film-



Where  $r$  is the circle's radius and  $l$  are the distance measured from the surface of the paper to the nozzle. The film-forming solution is more difficult to disseminate when sprayed at greater spray angles <sup>[12]</sup>.

### Drug Content per each spray actuation

The volume or amount of solution upon each spray is measured to establish drug content, and the concentration of the film-forming solution is used to quantify the dose uniformity. It is also possible to collect the sprayed solution and use

instruments to determine the concentration of the active ingredient <sup>[49]</sup>.

In order to get an accurate reading of the volume of spray that is produced, it is necessary to measure the total amount of solution that is contained within the sprayer rather than the weight of the film-forming solution that is expelled through the nozzle. Due to the fact that the spray droplets are so small and are easily carried by the wind, it is highly unlikely that all of the spray will be collected in order to be weighed. For calculating spray volume, use the formula below <sup>[14]</sup>:

$$V = \frac{W_t - W_o}{D} \quad \text{Eq. 2}$$

Where  $V$  represents the volume of spray that is produced with each actuation,  $W_t$  represents the weight of the spray solution after it has been sprayed,  $W_o$  represents the weight of the spray solution before it has been sprayed, and  $D$  represents the density of the solution <sup>[14]</sup>.

When working with metered-dose sprays, this test is absolutely necessary.

### In vitro Drug Release Study

Franz diffusion cells or USP dissolution apparatus type II (paddle Method), are typically employed in this test as compartment separators, along with cellulose membranes (pore size 0.44  $\mu\text{m}$ ), nylon membranes (particle size 0.21 mm), or silicone membranes <sup>[62]</sup>. A phosphate buffer with a pH of 6.8 is the medium <sup>[68]</sup>. After the system for the compartments has been established, the solution for making the film is poured into the donor compartment. At certain intervals of time, a quantity of the solution that is being measured is collected from the cells, and the equipment that is being used to measure it is then adjusted accordingly. After samples have been taken, they are followed by the replacement of the exact same volume of fluid <sup>[69]</sup>. After that, the drawn samples were examined with a UV visible spectrometer.

### Conclusion

The buccal mucosa is a suitable route to deliver drugs that cannot be absorbed via the digestive system (due to acidic medium of stomach, intestinal enzymes, or a strong hepatic first pass effect). It can also be inhaled or injected instead of being administered topically, intravenously, or nasally. The buccal mucosa's physiology permits the penetration of active compounds, and because of its quick cellular cycle and regeneration, penetration enhancers can be used. The latest studies have also demonstrated that adding permeability enhancers to buccal films did not impede production or cause mucosal irritation or toxicity. The design of buccal films still has a lot of possibility, including their recent application as system for the administration of nanoparticles and macromolecules to produce patient-friendly and safe dosage forms

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